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Groundwater Supply

Total Global Groundwater Volume = 23×10^{6} kilometers (6 x 10^{18} gallons)

Of that, 0.35 x 106 kilometers (9 x 1010 gallons) is < 50 years old $(1\frac{1}{2}\%)$

Science Daily, 16 Nov, 2015

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Domestic Well Water

- Supplies 15% of the U.S. population
- Not covered by U.S. EPA Safe Drinking Water Regulations

Water-Borne Contaminants

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Water Contaminants

Class	Examples
Suspended solids	Dirt, clay, colloidal materials, silt, dust, insoluble metal oxides and hydroxides
Dissolved organics	Trihalomethanes, synthetic organic chemicals, humic acids, fulvic acids
Dissolved ionics (salts)	Heavy metals, silica, arsenic, nitrate chlorides, sulfates
Microorganisms	Bacteria, viruses, protozoan cysts, fungi, algae, molds, yeast cells
Gases	Hydrogen sulfide, methane, radon, carbon dioxide





U.S. EPA lists drinking water contaminants by two standards:

- Primary Health Related
- Secondary Mainly Aesthetic Related
- Maximum Contaminant Level (MCL).
- \bullet Usually in mg/L (ppm) or μ g/L (ppb).
- New MCLs starting to be listed in ppt concentrations (1 second in ~ 32,000 years).



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Primary drinking water standards constantly evolving – virtually always getting tighter

A Case and Point -Manganese

- Used to be considered only aesthetic issue (staining).
- EPA Secondary Standard 0.05 ppm (mg/L)
- Now health concerns (MCL not determined)



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USGS Study of Domestic Wells (1991-2004) (Circular 1332)

- · 2100 private wells
- 48 states
- · 30 regionally extensive aquifers



- 23% at least one contaminant above MCL U.S. EPA Primary Drinking Water Standard
- 60% organic compounds (herbicides, pesticides, etc.) Low concentrations (PPCPs)
- 34% coliform bacteria

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 48% - at least one contaminant above MCL – U.S. EPA Secondary Drinking Water Standard

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Inorganic Ions

- Sodium
- Calcium
- Magnesium
- Strontium
- Boron

Iron

NitrateChloride

• Arsenic

• Uranium

• Fluoride

- Sulfate
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- Manganese

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Organic Compounds

Pesticides

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- Herbicides
- Volatile organics (benzene, chloroform, etc.)
- Tannins

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• Fulvic Acid





NGWA – currently 12 **BSP** documents -**Best Suggested Practices** – addressing waterborne contaminants

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These Cover:

- Iron and Manganese
- Strontium
- Boron
- Arsenic
- Uranium

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- Nitrate
- Radon
- · Methane
- · Hydrogen Sulfide
- Microorganisms
- Fluoride
- · PFAS (new)
- Perchlorate



Arsenic

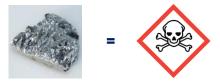
· Naturally occurring

· Hotspots:

- New England
- Upper Midwest
- South West
- Southern Texas
- ~ 2 million people exposed to high levels



Lead is a metal that occurs naturally in the environment



It is poisonous to humans and animals when consumed.

The presence of lead in groundwater tends to be very small almost undetectable.

A greater concern is the presence of lead in:

- Galvanized steel pipes
- Certain brass used in plumbing fittings and fixtures, or well components
- Certain solder used to connect pipes and joints

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Microorganisms





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Categories of Microorganisms

- Bacteria
- Protozoa
- Viruses

Microorganisms that can cause disease in humans are known as PATHOGENS

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Microorganisms

(CDC 2011-12) >50% of water-borne outbreaks from groundwater

PPCPs

Minnesota Department of Health Virus Study

- Similar occurrences between community and private wells
- 36+% system occurrence rate for the enteric viruses
- Wells previously thought not vulnerable appear to be susceptible to virus contamination

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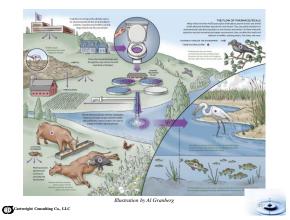
Also Known As:

- CECs (contaminants of emerging concern)
- EPOCs (emerging pollutants of concern)
- EPPPs (environmental persistent pharmaceutical pollutants)
- APIs (active pharmaceutical ingredients)
- EDCs (endocrine disrupting chemicals [compounds])



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PPCPs (Pharmaceutical and Personal Care Products)

- Unmetabolized drugs (in urine, feces)
- Discarded pharmaceutical products
- Cosmetics

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- Virtually every chemical flushed down the drain
- mostly organic and non-biodegradable

EDCs (Endocrine Disrupting Chemicals)

Agents that interfere with the synthesis, secretion, transport, binding or elimination of natural hormones in the body Responsible for maintenance of homeostasis, reproduction, development and behavior

PPCPs Are Mostly

Organic Chemicals

Estimated 80,000-100,000

• Virtually all human-made

Concentrations in Drinking Water

parts per trillion (ppt)

1 ppt =

1 second in ~32,000 years 6" leap in journey to the sun Pinch of salt in 10,000 tons of potato chips

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Perfluorinated Compounds (PFCs)

- Also known as PFAS polyfluoroalkyl substances
- Large family of fluorine containing organic compounds
- Very resistant to biodegradation
- Affect immune system development in fetuses





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Examples

- Perflurooctanic Acid (PFOA) Teflon[®]
- Perfluorooctane Sulfonate (PFOS) -Scotchgard[®]

New NGWA BSP

"Groundwater and PFAS: State of Knowledge and Practice"

PPCPs

- 1) Are the concentrations increasing?
- 2) Do the current concentrations have any effect on human health?

As population increases, PPCP concentrations increasing

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Are any of these contaminants in such tiny concentrations a risk?



- Last 30 years, birthrate for males declined every year
- Reproductive organ abnormalities increased 200% over last 20 years
- 85% of sperm DNA in healthy males damaged
- 300% increase in testicular cancer



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Insoluble materials generally in size range of nanometers (10⁻⁹ meter)



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Also Known As:

- Microfibers
- Nanofibers
- Microplastics
- Nanomaterials
- Nanoplastics



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Nanofibers

- Fabric finishes
- •Antifouling paints
- Degradation byproducts



- Cosmetic beads
- Lint

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Fleece fragments





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Yet to be **Determined:**

- · Long term effects of exposure to PPCPs & particles
- · New chemicals resulting from mixtures of PPCPs
- · Effects on the immunocompromised
- · PPCP effects on bacteria, fungal and aquatic life
- · Effects on antibiotic resistance

No Link of PPCPs and Nanoparticles to Human Health YET





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- Many studies underway
- No link identified yet
- Opinion only a matter of time

Treatment for Contaminant **Reduction**

No Single Treatment Technology

Multibarrier approach is required

Optimum Technology Depends Upon:

- · Contaminant Chemistry
- Concentrations
- Potential Effect (Health, Aesthetic, etc.)
- Degree of Removal Requirement
- Economics

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Treatment Locations

POE – Point of Entry **POU** – Point of Use

Treatment Summary

Contaminant	POE TREATMENT CHOICES			POU TREATMENT CHOICES		
Iron (Fe)	Oxidation/Filtration	IX	Softening	None		
Manganese (Mn)	Oxidation/Filtration	IX	Softening	None		
Strontium (Sr)	RO/Filtration	IX		RO	IX	Distillation
Lead (Pb)	RO	IX	Replace Components	RO	IX	Distillation
Boron (B)	RO/Filtration	IX		RO	IX	Distillation
Arsenic (As)	RO*	IX*	Adsorptive Media	RO*	IX*	Adsorptive Media
Uranium (U)	RO	IX		RO	IX	Distillation
Fluoride (F)	RO	IX	Adsorptive Media	RO	IX	Distillation
Nitrate (NO ₃)	RO	IX		RO	IX	Distillation
Perchlorate (ClO ₄)	RO	IX		RO	IX	Distillation
Radon (Rn)	Aeration					
Methane (CH ₄)	Aeration					
Hydrogen Sulfide (H2S)	Aeration	AC	Oxidation		AC	Oxidation
Microorganisms	Disinfection			Ozonation	UF	UV
* Following oxidation						

IX = Ion Exchange

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PPCP Removal Technologies

Treatment Technologies	Molecular Weight of Organic (Daltons)	Other Removal Characteristics
Activated Carbon		Primarily, more aromatic organics are adsorbed
Microfiltration	>100,000	Removes most particles
Ultrafiltration	>10,000	Removes most particles
Nanofiltration	>400	Virtually all particles and most multivalent ionic salts
Reverse Osmosis	>100	Almost all ionic salts removed as well as particles

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Adsorption with Activated Carbon

Low Molecular Weight Relatively Volatile Organics





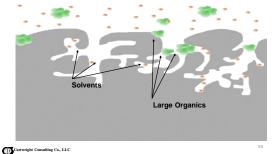


- Absorption a substance penetrating into the carbon
- Adsorption a substance adhering to the surface of the carbon





Activated Carbon Structure

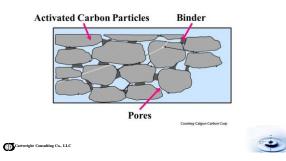


Carbon Types

Bituminous Lignite Wood Coconut Shell

Carbon Block

1,000 x magnification



Membrane Technologies

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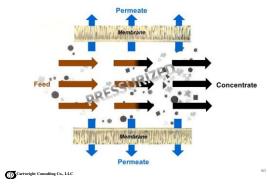


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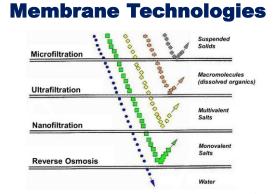
Membrane Filtration

Microfiltration Ultrafiltration Nanofiltration Reverse Osmosis



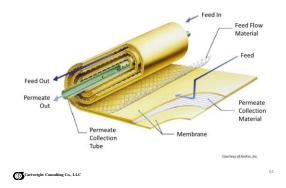






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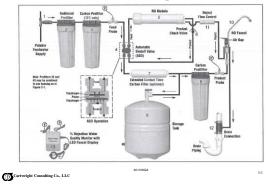
Spiral Wound Membrane



POU Reverse Osmosis Unit Components



POU RO System



Normal POU RO System

- Softening (hardness)
- Filtration (sediment)
- GAC (chlorine) for TFC membranes

Advanced Oxidation Processes (AOPs)

Ozone Ozone/H₂O₂ Ozone/UV UV/TiO₂ UV/H₂O₂ Hydroxyl Radical (•OH)

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- Break Chemical Bonds
- Form More Easily Treated (or less harmful) Compounds



- Best disinfectant
- Strong oxidant
- Destroys biofilms (slowly)
- Dissipates, converts to O₂
- Very little rinsing required
- Test strips or test kit to monitor

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Ion Exchange Technology

- Uses resins to adsorb ions (lead, manganese, hardness, etc.)
- Regenerated onsite

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